



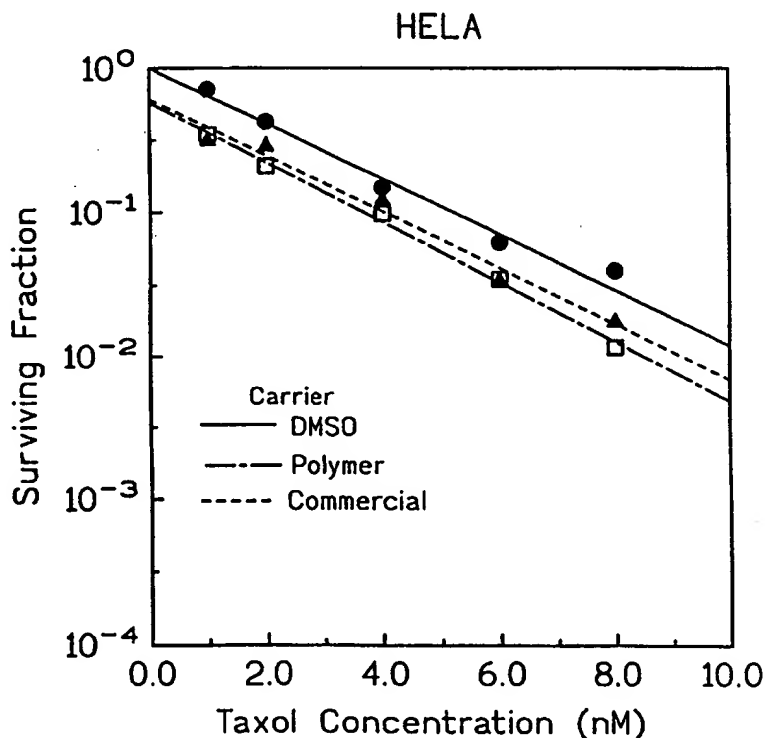
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(54) Title: DELIVERY SYSTEM

(57) Abstract

The present invention relates to a delivery system. More particularly it relates to a method of packaging a water-insoluble substance, such as, for example, a drug or other therapeutic or diagnostic agent, to facilitate uptake into the human or animal body and to the so packaged substance. The water solubilized form of a water-insoluble drug or other therapeutic or diagnostic agent is characterised in that the water-insoluble drug or other therapeutic or diagnostic agent is packaged in an amphiphilic polymer which is miscible in water. The invention also relates to the use of an amphiphilic polymer in the manufacture of a medicament.



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DESCRIPTIONDELIVERY SYSTEM

The present invention relates to a delivery system. More particularly it relates to a method of packaging a water-insoluble substance, such as, for example, a drug or other therapeutic or diagnostic agent, to facilitate uptake into the human or animal body and to the so packaged substance. It also relates to the use of an amphiphilic polymer in the manufacture of a medicament.

Many substances, both natural and manmade, whilst having potential as drugs or other therapeutic or diagnostic agents are rejected in early studies because they are water-insoluble both in in-vitro cell culture and subsequent in-vivo phase I clinical studies.

Current attempts to deliver water-insoluble substances in medicaments mostly involve some form of encapsulation (not an ideal solution because of the complex chemistry involved in getting uniform release of the active substance), micelle or similar formation using lipid-like materials (not ideal because each

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active substance needs its own special formulation and the micelles are dynamic systems that open and close regularly so that the agent is lost prematurely) or use of a water-miscible solvent such as dimethyl sulphoxide (not a desired solution because of the need to inject or apply solvent into human metabolic systems).

It is an aim of the present invention to produce a water-solubilized form of a water-insoluble substance.

According to one aspect of the present invention there is provided a water-solubilized form of a water insoluble drug or other therapeutic or diagnostic agent characterised in that the water insoluble drug or other therapeutic or diagnostic agent is packaged in an amphiphilic polymer which is miscible with water.

The drug or other therapeutic or diagnostic agent is carried internally within a hydrophobic or lipophilic pocket or pockets defined by the conformation of said polymer in the aqueous environment.

Thus, in an aqueous system, the water-insoluble substance is carried in the hydrophobic or lipophilic pocket or pockets of the amphiphilic polymer, the

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latter (and therefore the whole system) being soluble in water. On reaching for example a cell membrane, the amphiphilic polymer cannot cross the membrane because of its size but it can undergo a conformational change to release its contents, namely the water-insoluble substance, to the cell wall.

The amphiphilic polymer is preferably a copolymer comprising hydrophilic monomers and hydrophobic monomers.

An advantage of the amphiphilic polymer is its ability per macromolecule to carry several molecules of the drug or other therapeutic or diagnostic agent (often about 5 to 20) internally within the hydrophobic or lipophilic pocket. This means less amphiphilic polymer is required per unit of drug or other therapeutic or diagnostic agent than other carriers.

A further advantage of the amphiphilic polymer is that it allows attachment of cell or tissue specific molecules such as, for example, antibodies. The attachment of such molecules provides a means of targeting drugs contained within the amphiphilic polymer. In addition, a major problem with therapeutic

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antibody targeting at present is that the relatively small number of antigenic sites on cells restricts the maximum concentration of drug or other therapeutic agent that can be delivered to the cell. By labeling the amphiphilic polymer it should be possible to deliver much higher concentrations of the drug or other therapeutic or diagnostic agent to the cell. By attaching the amphiphilic polymer to, for example, an antibody, it would be possible to target tumour or other cells with high concentrations of the water-insoluble drug by presenting them to the tumour or other cells using the amphiphilic polymer as a carrier.

Another characteristic of the amphiphilic polymers lie in their electronic charge, which means in solution they can flow under the influence of an electrical potential gradient. Thus, with the drug or other therapeutic or diagnostic agent carried within a pocket or pockets defined by the conformation of said polymer in the aqueous environment the packaged drug or other therapeutic or diagnostic agent can be moved through a tissue under the influence of an electrical potential gradient. This ability would be advantageous in

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techniques in which molecules are transported from the surface of the skin to deeper layers by an electrical gradient.

Preferred amphiphilic polymers include:

polymers comprising at least one hydrophilic group and at least one hydrophobic group, such that the ratio of hydrophilic groups to hydrophobic groups lies in the range from 1:10 to 10:1; preferably from 1:3 to 3:1.

More preferably the amphiphilic polymers will comprise regions which are hydrophobic and regions which are hydrophilic in nature. As a consequence of the polymer having regions of differing hydrophobicity/hydrophilicity the polymer takes up a confirmation in an aqueous environment in which the hydrophobic regions are generally presented internally and the hydrophilic regions are generally presented externally. This results in the formation of a hydrophobic or lipophilic pocket or pockets which can be used to carry water-insoluble drugs or other therapeutic or diagnostic agents.

The hydrophobic groups or regions interact with the water insoluble drug or other therapeutic or

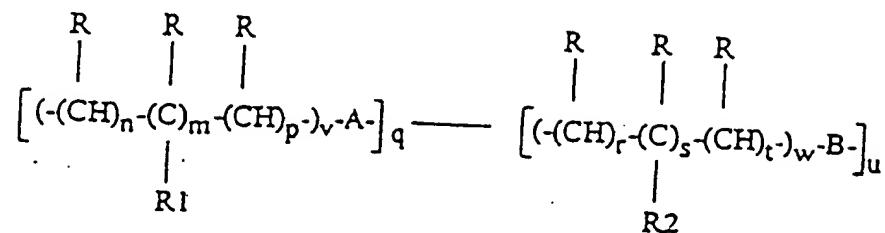
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diagnostic agent through hydrophobic non-bonded interactions such as dispersion forces, π - π interactions, and/or charge transfer interactions, and/or Van der Waals interactions which stabilise the composition in a pocket or pockets while the hydrophilic groups of the polymer solubilise the composition. When acting by charge transfer interactions, an electron donor-acceptor complex is formed.

Preferred polymeric structures for solubilisation of a drug or other therapeutic or diagnostic agent include the following:

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Structures of formula I.



where A and B are selected from :CH₂, NH, O, ketone, an ester linkage, an amide linkage, and an imine linkage;

where n, m, p, r, s, t can independently be any whole number as long as (n+m+p)_v + (r+s+t)_w ranges from 1-1000, preferably from 1-500 and such that the ratio v/w varies between 0.1 and 1;

where the values of q and u, independently lie in the range from 1-10;

Each R, independently, can be selected from H, alkyl, haloalkyl, alkenyl, and alkynyl; preferably H or CH₃;

R1 is selected from (partially) hydrophobic moieties containing aromatic hydrocarbon rings derived from compounds such as toluene, methyl styrene, stilbene, pyridine, naphthalene, anthracene,

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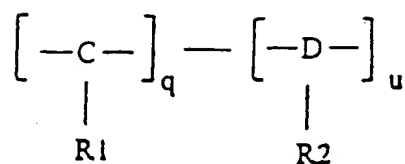
phenanthrene, phenyl, histidine, tryptophan, phenyl alanine, tyrosine, alkyl benzene, xylenes, carbazoles, xanthenes, acridines, purines, pyridazines and indoles;

R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature such as -OH, hydroxylalkyl such as hydroxymethyl and hydroxyethyl; polyoxyethylene, hydroxyphenyl and derivatives thereof, moieties derived from pyrrolidine, pyridine-N-oxide, N-oxide derivatives of histidine, tryptophan, phenyl alanine, tyrosine; phenyl sulphonate, naphthalene sulphonate, imidazole, water-soluble salt derivative of naphthalene, anthracene, phenanthrene, phenyl, carbazoles xanthenes, acridines, purines, pyridazines, indoles,, -COOH, -COOM, -SO₃M where M is an alkali metal ion ; -NR₂, -NR₃⁺X⁻ where X is a halide ion and R is, independently selected from H, alkyl, and hydroxyalkyl;

R1 and R2, independently, can be substituents on the moieties A and B in the event that A and/or B are not an ether, ketone, amide, imine, or ester linkage.

Substituted polysaccharides of the unit structure II.

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where C and D are oligosaccharide units and where oligosaccharide units comprise the product of polycondensation of monosaccharides by O- glycosidic linkage containing up to 10 such residues selected from hexose, pentose and deoxyhexose residues;

where the values of q and u, independently, lie in the range from 1 to 10;

where the substituent R1 is selected from (partially) hydrophobic moieties containing aromatic hydrocarbon rings derived from compounds such as toluene, methyl styrene, stilbene, pyridine, naphthalene, anthracene, phenanthrene, phenyl, histidine, tryptophan, phenyl alanine, tyrosine, alkyl benzene, xylenes, carbazoles, xanthenes, acridines, purines, pyridazines and indoles; and

R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature such as -OH, hydroxylalkyl such as hydroxymethyl and

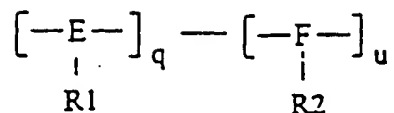
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hydroxyethyl; polyoxyethylene, hydroxyphenyl and derivatives thereof, moieties derived from pyrrolidine, pyridine-N-oxide, N-oxide derivatives of histidine, tryptophan, phenyl alanine, tyrosine; phenyl sulphonate, naphthalene sulphonate, imidazole, water-soluble salt derivative of naphthalene, anthracene, phenanthrene, phenyl, carbazoles xanthenes, acridines, purines, pyridazines, indoles,, -COOH, -COOM, -SO₃M where M is an alkali metal ion ; -NR₂, -NR₃⁺X⁻ where X is a halide ion and R is, independently selected from H, alkyl, and hydroxyalkyl;

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Polymers having a unit structure of the formula

III.



where E and F are di-substituted aromatic moieties such as phenylene, naphthalene, phenanthroline, anthracene, thiophene;

where the value of q and u, independently, lie in the range from 1 to 10;

R1 is selected from (partially) hydrophobic moieties containing aromatic hydrocarbon rings derived from compounds such as toluene, methyl styrene, stilbene, pyridine, naphthalene, anthracene, phenanthrene, phenyl, histidine, tryptophan, phenyl alanine, tyrosine, alkyl benzene, xylenes, carbazoles, xanthenes, acridines, purines, pyridazines and indoles; and

R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature such as -OH, hydroxylalkyl such as hydroxymethyl and hydroxyethyl; polyoxyethylene, hydroxyphenyl and

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derivatives thereof, moieties derived from pyrrolidine, pyridine-N-oxide, N-oxide derivatives of histidine, tryptophan, phenyl alanine, tyrosine; phenyl sulphonate, naphthalene sulphonate, imidazole, water-soluble salt derivative of naphthalene, anthracene, phenanthrene, phenyl, carbazoles xanthenes, acridines, purines, pyridazines, indoles,, -COOH, -COOM, -SO₃M where M is an alkali metal ion ; -NR₂, -NR₃⁺X⁻ where X is a halide ion and R is, independently selected from H, alkyl, and hydroxyalkyl;

The molecular weight of the amphiphilic oligomer or polymer is preferably in the range of from 500 to 1,000,000 D more preferably from 1000 to 500,000 D and most preferably 30,000 to 150,000 D.

The amphiphilic polymers described above can be random, graft, or block polymers.

The amphiphilic polymers can also contain more than two different monomers as long as the ratio of hydrophilic and hydrophobic groups lies in the range from 1:10 to 10:1; preferably from 1:3 : 3:1.

Preferred amphiphilic polymers are aromatic amphiphilic polymers selected from : polystyrene sodium

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sulphonate-co-vinyl naphthalene, polyvinyl pyrrolidone-co-vinyl imidazole, and polynaphthalene sulphonate (TAMOL®).

A more preferred copolymer is polystyrene sodium sulphonate-co-vinyl naphthalene.

Furthermore, the polymer can be a homopolymer which exhibits amphiphilic character such as partially neutralised polymethacrylic acid.

There is also another special group of saccharides that are suitable for the solubilisation of a drug or other therapeutic or diagnostic agent which are the cyclic polysaccharides such as natural cyclodextrins and modified cyclodextrins.

According to a further aspect of the present invention there is provided the use of an amphiphilic polymer in the manufacture of a medicament.

In one embodiment the therapeutic substances are those used in inhalants. The amphiphilic polymers of the invention could thus provide novel vehicles for solubilizing and delivering therapeutic substances to the mucosal lining of the lungs in inhalants, nebulisers etc. such as those used in the treatment of

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asthma.

It is another aim of the present invention to provide a way of packaging water-insoluble substances in order that they might be used as medicaments.

According to this aspect of the present invention there is provided a method of solubilising a water-insoluble drug or other therapeutic or diagnostic agent comprising packaging said drug or other therapeutic or diagnostic agent in an amphiphilic polymer which is miscible in water.

The drug or other therapeutic or diagnostic agent may be packaged as a water solubilized medicament or diagnostic tool using, for example, one of the following methods:

Preparation of water-soluble medicament

The process for making the water-soluble medicament comprises the following steps:

- a-dissolving the drug or other therapeutic or diagnostic agent in a water miscible organic material;
- b-dissolving said polymer in H₂O;
- c-adding said predissolved polymer to the

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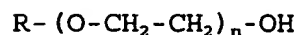
predissolved drug or other therapeutic or diagnostic agent or vice versa

d-adding as much water as may be necessary for dilution slowly with stirring; and

e-optionally removing the water miscible organic material by evaporation and the water by freeze drying.

The water miscible organic material may be any pharmaceutically acceptable material in which a drug or other therapeutic or diagnostic agent is soluble. A suitable material is a water-miscible organic solvent selected from methanol, ethanol, isopropanol, n-propanol, acetone, n-methylpyrrolidone and dimethylformamide, preferably methanol, ethanol, ethanol, isopropanol, n-propanol or acetone.

A particularly preferred water miscible material for the production of the water-soluble medicament in liquid form is a nonionic surfactant of formula



where R is a C₈-C₂₂ alkyl; and

where n lies in the range from 2 to 24, preferably 7.

Alternately a water-soluble medicament may be prepared by

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a) -Synthesis from solution or from a polymer melt or

b) -Isolation as a pure solid

a) -Synthesis of the water-soluble medicament

1. -From solution, the synthesis comprises the following steps:

i) the drug or other therapeutic or diagnostic agent is dissolved in a water miscible organic solvent, for example dimethylsulfoxide (DMSO) or dimethyl formamide (DMF).

ii) -dissolution or addition of the solid polymer to this solution, and

iii) -Addition of sufficient water to yield the required solution strength while the mixture is stirred continuously.

b-Isolation of the water-soluble medicament

1.-Isolation of the medicament as a pure solid. An aqueous solution of the amphiphilic polymer containing the therapeutic agent is mixed with an excess of acetone. During 2-4 hours a precipitate is formed which is isolated using suction filtration followed by vacuum drying. The resulting solid can be redissolved in water as needed.

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2. -From a polymer melt the synthesis comprises the following steps: the drug or other therapeutic or diagnostic agent is dissolved in molten polymer and stirred for half an hour. After cooling, the resulting solid can be added to water in order to prepare an aqueous solution of the water solubilized therapeutic agent.

3. - A solution of the therapeutic agent plus amphiphilic polymer together in a organic solvent is evaporated to dryness. The resulting solid may be dissolved in water and filtered if necessary to yield a water solubilized form of the therapeutic agent.

The amphiphilic polymers of the invention offer many potential medical applications 1) they could, for example, be used to test potentially therapeutically active drugs, 2) they could also provide an alternative vehicle for the delivery of some of the presently available drugs which require special techniques and reagents to make them soluble. 3) they might also be of value in carrying insoluble therapeutic molecules through the skin or to mucus membranes and 4) they might also allow insoluble

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agents with cancer specific localisation, ie. fluorinated porphyrins to be used in nmr spectroscopy and imaging of cancer.

The invention will be further described by way of example only with reference to the following example compositions and test data.

Example of delivery of a photochemical sensitizer to a culture of cancer cells (Molt 4).

(i) Preparation of an aqueous solution of a water-insoluble sensitizer.

Typically, between 0.3 and 0.5 g of an amphiphilic polymer eg. polystyrene sodium sulphonate co vinyl naphthalene and 2-4 mg of a sensitizer eg 5,10,15,20 tetrakis phenyl porphyrin were dissolved in approximately 0.5 mL of dimethylformamide by stirring the ingredients together for about 5 minutes at room temperature. The dimethylformamide was evaporated off under reduced pressure. To the residue was added 1 L of distilled water and the whole was stirred for 24 hours in the dark at about 50°C. After this period, the solution was filtered through a 0.22 μ m pore size filter (Millipore) to give a clear solution. The exact

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concentration of sensitizer in the amphiphilic polymer was determined by analysis. Typically, for a porphyrin sensitizer, the concentration was determined by ultraviolet/visible measurement of absorption at the Soret wavelength near 420nm.

(ii) Preparation of cell growth medium.

The cell growth medium was prepared in three varieties. Each variety contained the nutrients required for cell growth. One variety contained neither sensitizer nor polymer, a second contained polymer but no sensitizer and a third contained both polymer and sensitizer.

Into a flask were mixed RPMI (Roswell Park Memorial Institute) tissue culture medium; from Gibco-BRL; 10 mL at 10 times normal concentration, aqueous sodium bicarbonate solution (3 mL of 7.5% w/v), L-glutamine (1 mL; 2 mM), penicillin and streptomycin (1 mL; 50 μ mg/mL) and the sensitizer solution prepared as in (i) above (85 mL). The resulting solution was adjusted to pH 7.2 by careful addition of aqueous sodium hydroxide solution (5 M) and finally once more filtered through a 0.22 μ m filter. The resulting

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solution was used as the growth medium containing both sensitizer and polymer in the following experiments. A growth medium containing only polymer was prepared as here but omitting the sensitizer, viz, instead of the sensitizer solution (85 mL), an equal volume of distilled water containing the same amount of polymer but no sensitizer was added. For the growth medium containing no sensitizer and no polymer, only distilled water (85 mL) was used in the above formulation in place of the sensitizer solution.

(iii) Testing the effect of the "water-insoluble" sensitizer held in aqueous solution by the amphiphilic polymer.

For this testing, three series of solutions were prepared. For any one concentration of a particular sensitizer, the three solutions were:

(a) A control, consisting of only the growth medium with no sensitizer and no polymer present.

(b) A solution containing the growth medium plus polymer but with no sensitizer present.

(c) A solution containing the growth medium plus polymer plus sensitizer.

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For each growth medium, 10 mL of a suspension containing 10^7 - 10^8 cells/mL were centrifuged and the cells were resuspended in one of the growth media described above (45mL) to give three cell preparations: A, containing sensitizer plus polymer; B, containing neither sensitizer nor polymer; C, containing polymer but no sensitizer. Serum (5 mL) was added to each flask and the cells were incubated at 37°C for 24 hours. Thus during cultivation, one set of cells was exposed to sensitizer plus polymer, one set to neither sensitizer nor polymer and a third set to polymer but not sensitizer. The flasks A,B,C were taken from the incubator. A cell count showed no significant difference in the numbers of cells in each growth medium. Thus, cell growth was not inhibited by either the sensitizer or the polymer. The contents of flask A containing both sensitizer and polymer were divided into two approximately equal parts (A_1 , A_2). Part A_1 was reserved. Part A_2 was centrifuged down and the supernatant liquid medium was removed; the cells were washed three times by resuspending them in Dulbecco's phosphate buffered saline (PBS;pH 7) and centrifuging

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them down again. This procedure removed any sensitizer and polymer from the exterior of the cells and left only sensitizer on the inside of the cells. Its amount could be determined by visible spectroscopy as described above. The washed cells were finally resuspended in fresh growth medium containing no sensitizer and no polymer (25mL). In the same way, the contents of flask B were divided into two equal portions (B_1 , B_2). Part B_1 was reserved and part B_2 was washed with PBS as described for part A_2 and then resuspended in fresh growth medium containing neither polymer nor sensitizer. Flask C was divided into two equal portions C_1 , C_2 . The viability of the cells in flask A_1 , A_2 , B_1 , B_2 , C_1 and C_2 was checked.

For irradiation experiments, the contents of the flasks A_1 , A_2 , B_2 and C_2 were irradiated with white light for one hour. All cell preparations, including the controls (B_1 , C_1) were incubated for a further 24 hours.

After a typical irradiation experiment, the cell viability was checked by counting the density of cells immediately after the 1 hour of irradiation and again

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24 hours after irradiation has been stopped. Typical results are shown in the Table.

Table showing cell viability after irradiation and incubation for 24 hours.

	A ₁	A ₂	B ₂	C ₂
Percentage of dead cells before irradiation ^a	4	4	3	5
Percentage of dead cells immediately after irradiation ^b	51	47	5	8
Percentage of dead cells 24 hours after irradiation	100	100	4	6

^a These were the percentages of dead cells after 24 hours of incubation in the absence of light.

^b These were the percentages of dead cells immediately after irradiation for 1 hour with white light.

^c These were the percentages of dead cells after a further 24 hours on incubation following the 1 hour irradiation. The observation of 5-10 dead cells per 100 is quite normal for typical cell preparations in which no extraneous factors are at work. The controls B₁ and C₁ showed proportions of dead cells of only 5-8%.

Example of delivery of a cytotoxic agent to cultures of cancer cells

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To Test the effects of cell cytotoxins.

Taxol is a powerful cytotoxic agent but is very difficult to use because it is almost insoluble in water or blood plasma. For *in vivo* or *in vitro* use, it is generally dissolved in dimethyl sulphoxide (DMSO) or in a complex commercial pharmaceutical preparation. For the experiments described here, Taxol was incorporated into an aqueous amphiphilic polymer solution exactly as described for the photosensitizer in part (i), except that the Taxol was included in place of the sensitizer. The Taxol concentration was varied over the range of 1 to 8nM. These carrier solutions were labelled "polymer" series. Similarly, the same range of Taxol solution concentrations was prepared in neat dimethyl sulphoxide as solvent; these carrier solutions were labelled "DMSO" series. Finally, a series of commercially available pharmaceutical preparations was used, which contained Taxol in the concentration range of 1 to 8 nM; these carrier solutions were labelled "commercial" series. Finally, a control series was used, in which only the amphiphilic polymer solutions were used but without

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containing any Taxol. The four series of solutions were compared for their cytotoxic action by a clonogenic cell survival assay using three human in vitro cell lines, viz., HELA (originally explanted from cancer of the cervix), HRT-18 (originally explanted from a lung cancer) and MGHU-1 (originally explanted from a bladder cancer). The results are shown graphically in Figs. 1 to 3.

Fig. 1 illustrates the effect of the polymer in the delivery of Taxol in a Hela cell line;

Fig. 2 illustrates the effect of the polymer in the delivery of Taxol in a HRT-18 cell line; and

Fig. 3 illustrates the effect of the polymer in the delivery of Taxol in a MGHU-1 cell line.

In the controls with no Taxol, the cells proliferated normally. At increasing concentrations of Taxol in the commercial and DMSO series, the survival of the three cancer cell lines decreased steadily until at 8 mM the surviving fractions were about 0.1 to 0.01 in HELA and HRT-18, compared with an initial fraction equal to 1.0 but were only 0.0005 in MGHU-1. The "commercial" preparations were somewhat more effective

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than were the "DMSO" series. The "polymer" series behaved almost identically to the commercial series for all three cell lines, showing that these aqueous amphiphilic polymer solutions were equally as effective as the commercial or DMSO preparations. The controls showed that aqueous solutions of the amphiphilic polymer without Taxol were inactive in cytotoxicity and therefore that the observed effect with Taxol was due entirely to the Taxol and not the carrier. The results of the assays are shown in the figures.

In a similar fashion, the water insoluble cytotoxic agent, amsacrine, was shown to be effective for cytotoxicity of the above cell lines when it was incorporated into aqueous solutions of an amphiphilic polymer.

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CLAIMS

1. A water solubilized form of a water-insoluble drug or other therapeutic or diagnostic agent characterised in that the water-insoluble drug or other therapeutic or diagnostic agent is packaged in an amphiphilic polymer which is miscible in water.

2. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in claim 1 wherein the water insoluble drug or other therapeutic or diagnostic agent is carried internally within a hydrophobic or lipophilic pocket or pockets defined by the conformation of said polymer in the aqueous environment.

3. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in claim 1 or 2 wherein the amphiphilic polymer comprises at least one hydrophilic group and at least one hydrophobic group, such that the ratio of hydrophilic groups to hydrophobic groups lies in the range from 1:10 to 10:1.

4. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as

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claimed in claim 3 wherein the ratio of hydrophilic groups to hydrophobic groups lies in the range from 1:3 to 3:1.

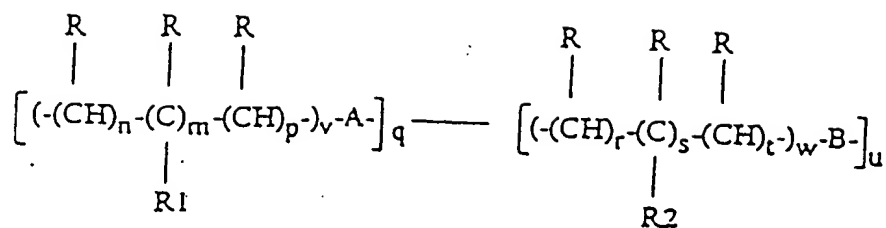
5. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the polymer is a copolymer comprising one or more hydrophilic monomers and one or more hydrophobic monomers.

6. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the water-insoluble drug or other therapeutic or diagnostic agent is packaged in an amount up from 5 to 20 molecules per macro molecule of amphiphilic polymer.

7. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the amphiphilic polymer is selected from the group consisting of:

Structures of formula I.

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where A and B are selected from :CH₂, NH, O, ketone, an ester linkage, an amide linkage, and an imine linkage;

where n,m,p,r,s,t can independently be any whole number as long as (n+m+p)_v+(r+s+t)_w ranges from 1-1000, and the ratio v/w varies between 0.1 and 1;

where the values of q and u, independently lie in the range from 1-10;

Each R, independently, can be selected from H, alkyl, haloalkyl, alkenyl, and alkynyl;

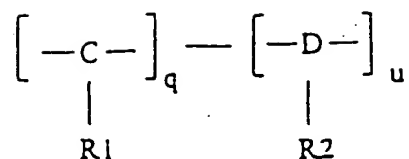
R1 is selected from (partially) hydrophobic moieties containing aromatic hydrocarbon rings;

R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature;

R1 and R2, independently, can be substituents on the moieties A and B in the event that A and/or B are not an ether, ketone, amide, imine, or ester linkage;

Substituted polysaccharides of the unit structure II.

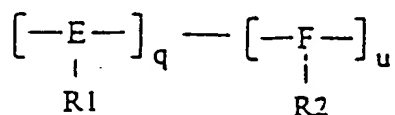
-30-



where C and D are oligosaccharide units and where oligosaccharide units comprise the product of polycondensation of monosaccharides by O- glycosidic linkage containing up to 10 such residues selected from hexose, pentose and deoxyhexose residues; where the values of q and u, independently, lie in the range from 1 to 10; and where the substituents R1 is selected from (partially) hydrophobic moieties containing aromatic hydrocarbon rings;

R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature;

Polymers of structure of the formula III.



where E and F are di-substituted aromatic moieties; where the value of q and u, independently, lie in the range from 1 to 10; R1 is selected from (partially)

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hydrophobic moieties containing aromatic hydrocarbon rings; and R2 is selected to provide water-solubility and these substituents are therefore hydrophilic in nature.

8. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in claim 7 wherein in the structures of formula 1, 2 and 3 the (partially) hydrophobic moieties containing aromatic hydrocarbon rings of R1 are derived from the group consisting of toluene, methyl styrene, stilbene, pyridine, naphthalene, anthracene, phenanthrene, phenyl, histidine, tryptophan, phenyl alanine, tyrosine, alkyl benzene, xylenes, carbazoles, xanthenes, acridines, purines, pyridazines and indoles; and the substituents R2 are selected from the group consisting of -OH, hydroxylalkyl; polyoxyethylene, hydroxyphenyl and derivatives thereof, moieties derived from pyrrolidine, pyridine-N-oxide, N-oxide derivatives of histidine, tryptophan, phenyl alanine, tyrosine; phenyl sulphonate, naphthalene sulphonate, imidazole, water-soluble salt derivative of naphthalene, anthracene, phenanthrene, phenyl, carbazoles xanthenes,

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acridines, purines, pyridazines, indoles,, -COOH, -COOM, -SO₃M where M is an alkali metal ion ; -NR₂, -NR₃⁺X where X is a halide ion and R is, independently selected from H, alkyl, and hydroxyalkyl; and in the structure of formula 3 E and F are selected from the group consisting of phenylene, naphthalene, phenanthroline, anthracene, thiophene;

9. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the amphiphilic polymer has a molecule weight of from 30,000 to 150,000 D.

10. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the amphiphilic polymer is a random, graft or a block polymer.

11. A water solubilized form of a water insoluble drug or other therapeutic or diagnostic agent as claimed in any of the preceding claims wherein the amphiphilic polymer is selected from polystyrene sodium sulphonate-co-vinyl naphthalene,

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polyvinyl pyrrolidone-co-vinyl imidazole,
polynaphthalene sulphonate (TAMOL®); polystyrene sodium
sulphonate-co-vinyl naphthalene;

partially neutralised polymethacrylic acid;

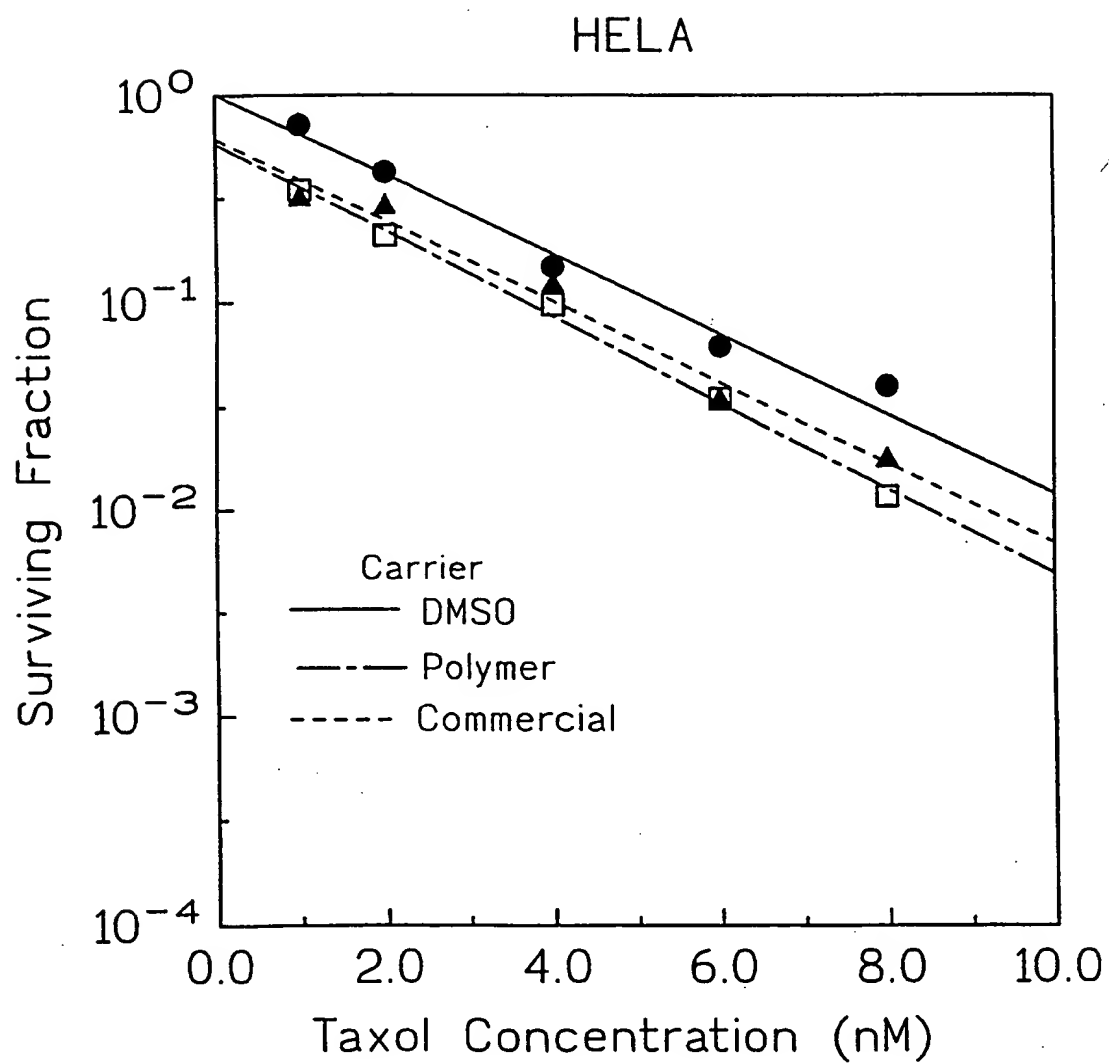
natural cyclodextrins and modified cyclodextrins.

12. A method of solubilising a water-insoluble
drug or other therapeutic or diagnostic agent
comprising packaging said drug or other therapeutic or
diagnostic agent in an amphiphilic polymer which is
miscible with water.

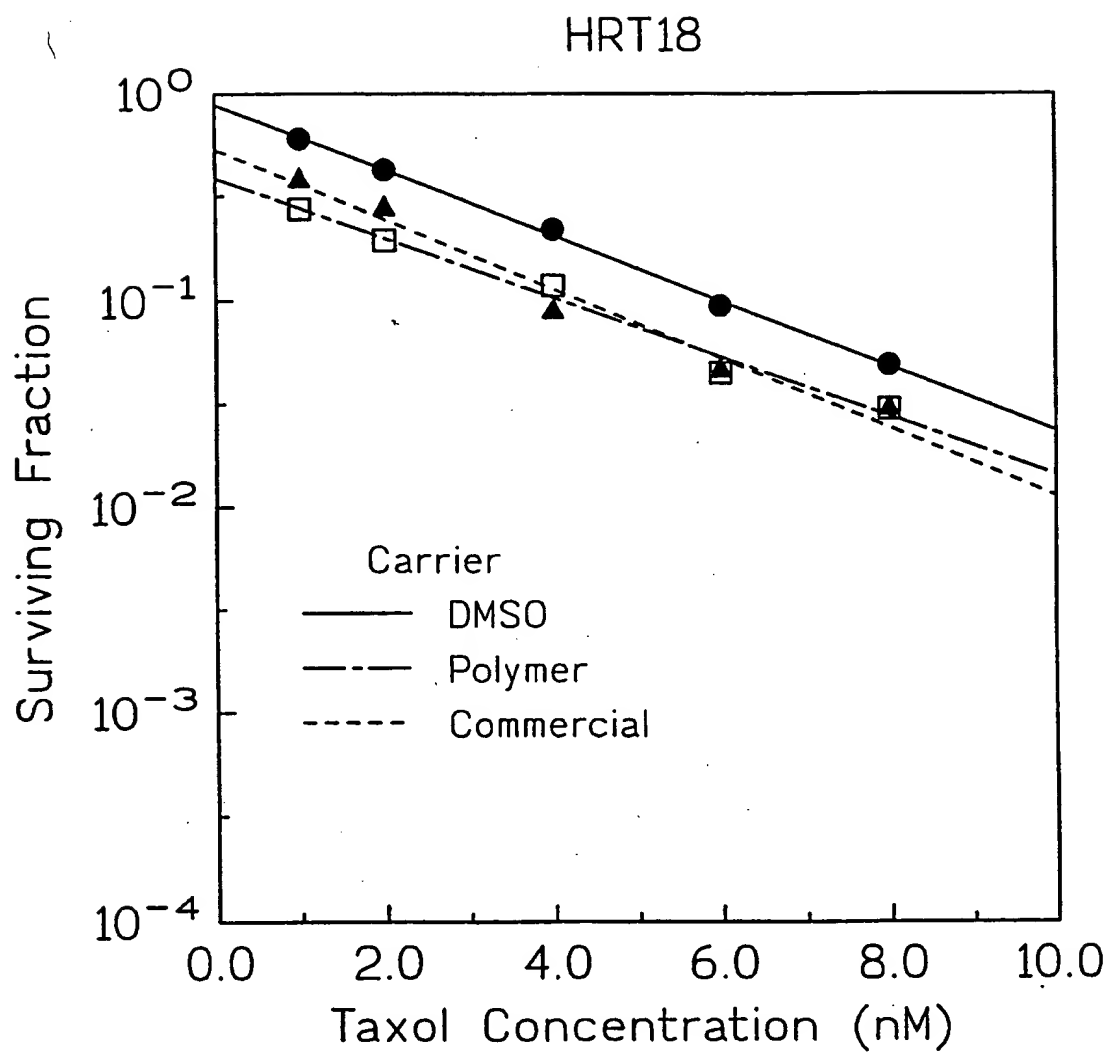
13. The use of an amphiphilic polymer in the
manufacture of a medicament.

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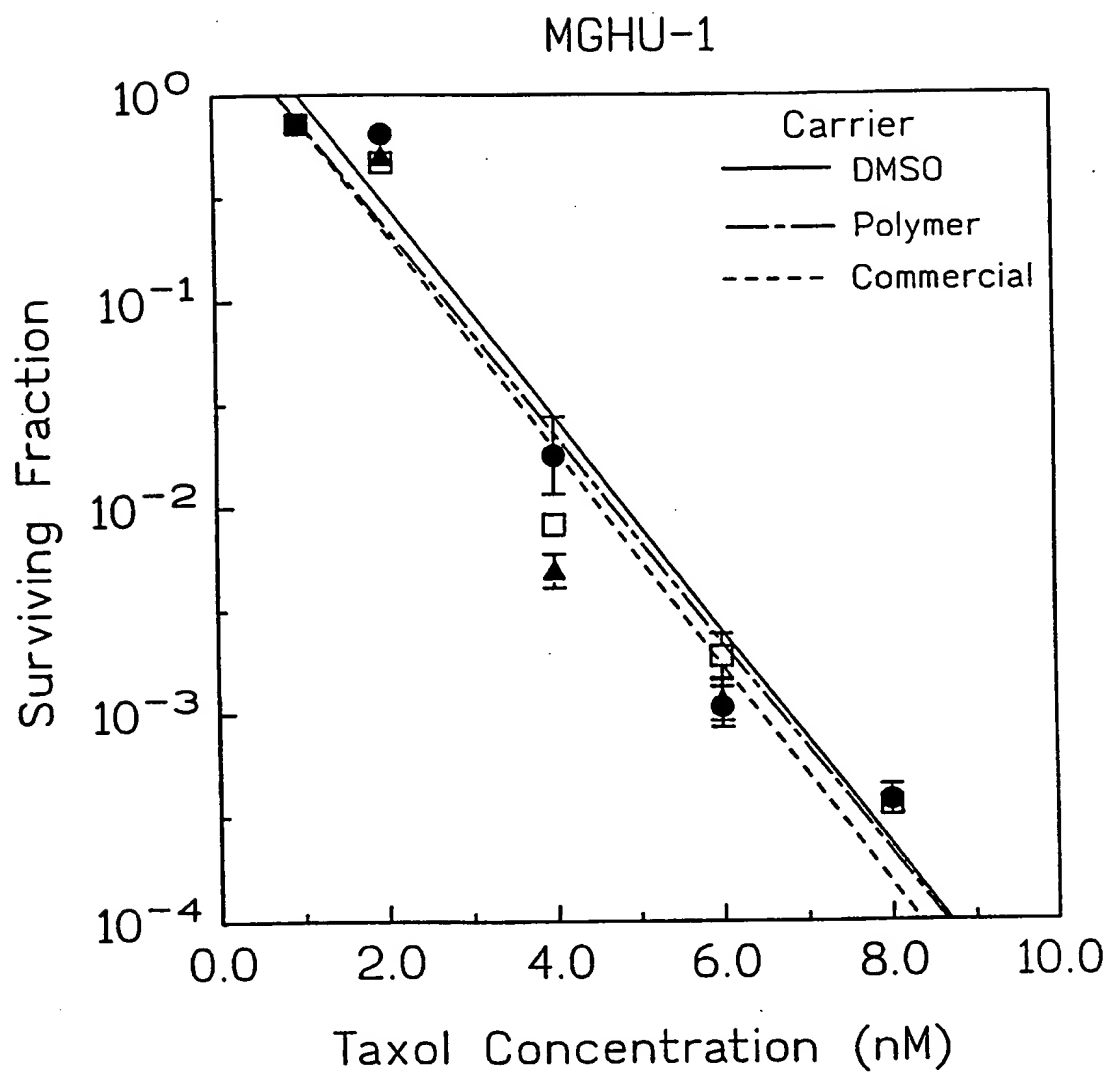
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**FIG. 1**

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**FIG. 2**

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**FIG. 3**



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : A61K 47/32		A3	(11) International Publication Number: WO 99/18998																												
			(43) International Publication Date: 22 April 1999 (22.04.99)																												
(21) International Application Number: PCT/GB98/03046 (22) International Filing Date: 9 October 1998 (09.10.98) (30) Priority Data: 9721367.2 9 October 1997 (09.10.97) GB (71) Applicant (for all designated States except US): THE UNIVERSITY OF LIVERPOOL [GB/GB]; Senate House, Abercromby Square, Liverpool L69 3BX (GB). (72) Inventors; and (75) Inventors/Applicants (for US only): JOHNSTONE, Robert, Alexander, Walker [GB/GB]; 39 Poulton Road, Bebington, Merseyside L63 9LD (GB). WARENIUS, Hilmar, Meek [GB/GB]; 14 Delavor Road, Heswall, Merseyside L64 RM (GB). FIGUEIREDO, Telmo, Luis, Coutinho [PT/PT]; Universidade de Aveiro, Dept. de Quimica, P-3800 Aveiro (PT). SØRENSEN, Alexandra, Maria, Pocas de Santana [DK/DK]; Vangede, Bygade 115 2th, DK-2820 Gentofte (DK). (74) Agent: W.P. THOMPSON & CO.; Coopers Building, Church Street, Liverpool L1 3AB (GB).		(81) Designated States: JP, US, European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments. (88) Date of publication of the international search report: 22 July 1999 (22.07.99)																													
(54) Title: DELIVERY SYSTEM																															
(57) Abstract																															
<p>The present invention relates to a delivery system. More particularly it relates to a method of packaging a water-insoluble substance, such as, for example, a drug or other therapeutic or diagnostic agent, to facilitate uptake into the human or animal body and to the so packaged substance. The water solubilized form of a water-insoluble drug or other therapeutic or diagnostic agent is characterised in that the water-insoluble drug or other therapeutic or diagnostic agent is packaged in an amphiphilic polymer which is miscible in water. The invention also relates to the use of an amphiphilic polymer in the manufacture of a medicament.</p>																															
<div style="text-align: center;"> <p>HELA</p> <table border="1"> <caption>Approximate data points from the HELA graph</caption> <thead> <tr> <th>Taxol Concentration (nM)</th> <th>DMSO (Surviving Fraction)</th> <th>Polymer (Surviving Fraction)</th> <th>Commercial (Surviving Fraction)</th> </tr> </thead> <tbody> <tr> <td>0.0</td> <td>1.0</td> <td>1.0</td> <td>1.0</td> </tr> <tr> <td>2.0</td> <td>0.5</td> <td>0.3</td> <td>0.3</td> </tr> <tr> <td>4.0</td> <td>0.2</td> <td>0.1</td> <td>0.1</td> </tr> <tr> <td>6.0</td> <td>0.1</td> <td>0.05</td> <td>0.05</td> </tr> <tr> <td>8.0</td> <td>0.05</td> <td>0.02</td> <td>0.02</td> </tr> <tr> <td>10.0</td> <td>0.02</td> <td>0.01</td> <td>0.01</td> </tr> </tbody> </table> </div>				Taxol Concentration (nM)	DMSO (Surviving Fraction)	Polymer (Surviving Fraction)	Commercial (Surviving Fraction)	0.0	1.0	1.0	1.0	2.0	0.5	0.3	0.3	4.0	0.2	0.1	0.1	6.0	0.1	0.05	0.05	8.0	0.05	0.02	0.02	10.0	0.02	0.01	0.01
Taxol Concentration (nM)	DMSO (Surviving Fraction)	Polymer (Surviving Fraction)	Commercial (Surviving Fraction)																												
0.0	1.0	1.0	1.0																												
2.0	0.5	0.3	0.3																												
4.0	0.2	0.1	0.1																												
6.0	0.1	0.05	0.05																												
8.0	0.05	0.02	0.02																												
10.0	0.02	0.01	0.01																												

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 98/03046

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 A61K47/32

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

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X	EP 0 583 955 A (RESEARCH DEVELOPMENT CORPORATION OF JAPAN) 23 February 1994 see the whole document ---	1-7,9, 10,12,13
X	WO 97 10849 A (SAM YANG CO., LTD.) 27 March 1997 see the whole document ---	1-7,9, 10,12,13
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Date of the actual completion of the international search

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Date of mailing of the international search report

28/05/1999

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International Application No

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